

# Reversible computing and the resource-theoretic approach to thermodynamics

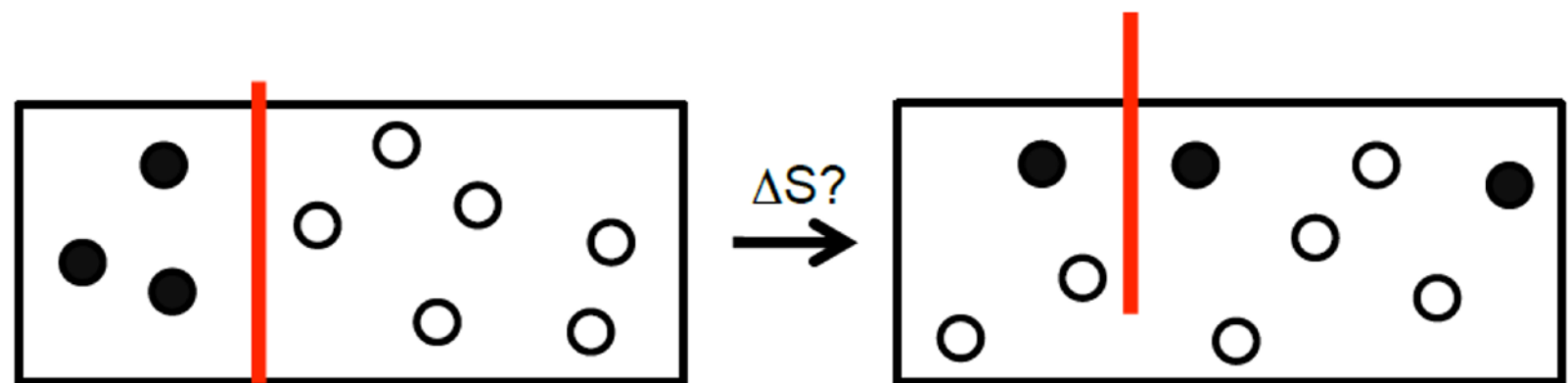
Markus P. Müller

<sup>1</sup> Institute for Quantum Optics and Quantum Information, Vienna

<sup>2</sup> Perimeter Institute for Theoretical Physics, Waterloo, Canada

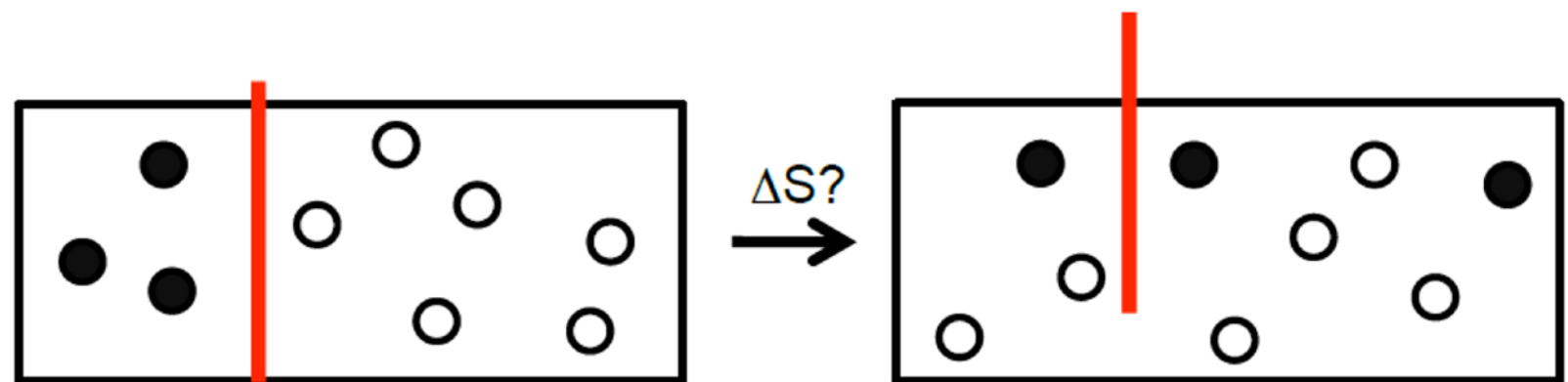


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Recall thermodynamics at **fixed background temperature  $T$** .



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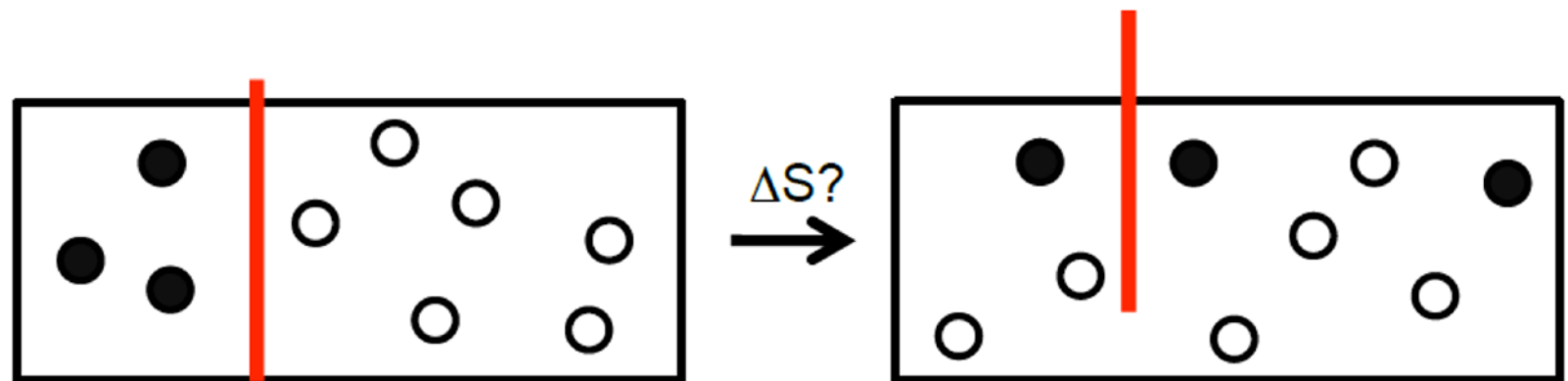
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$$\Delta F \leq 0 \quad (\text{2nd law}),$$

where  $F = U - TS$ .

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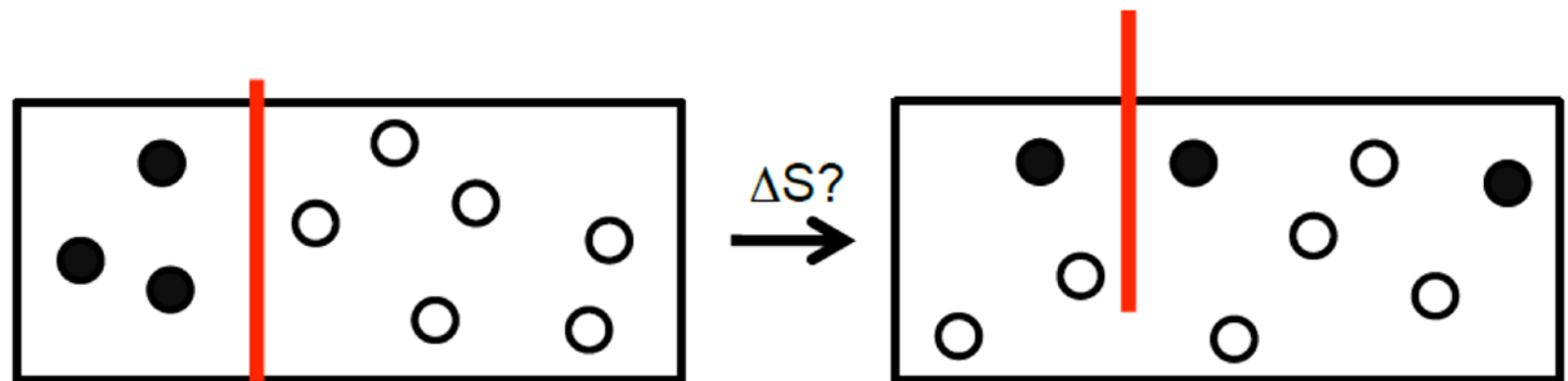
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But this is a statement **on average**, since “work” is a random variable.

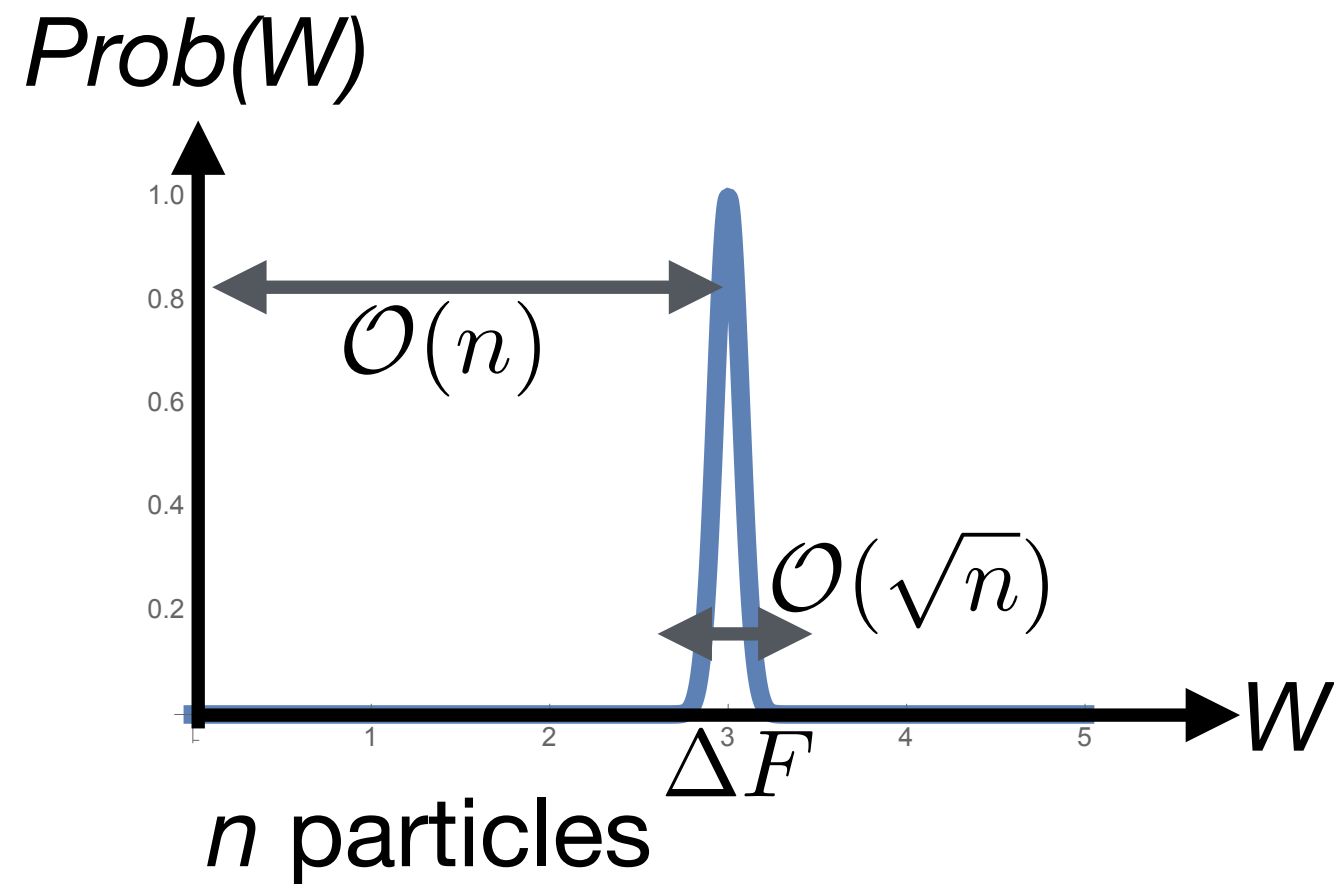
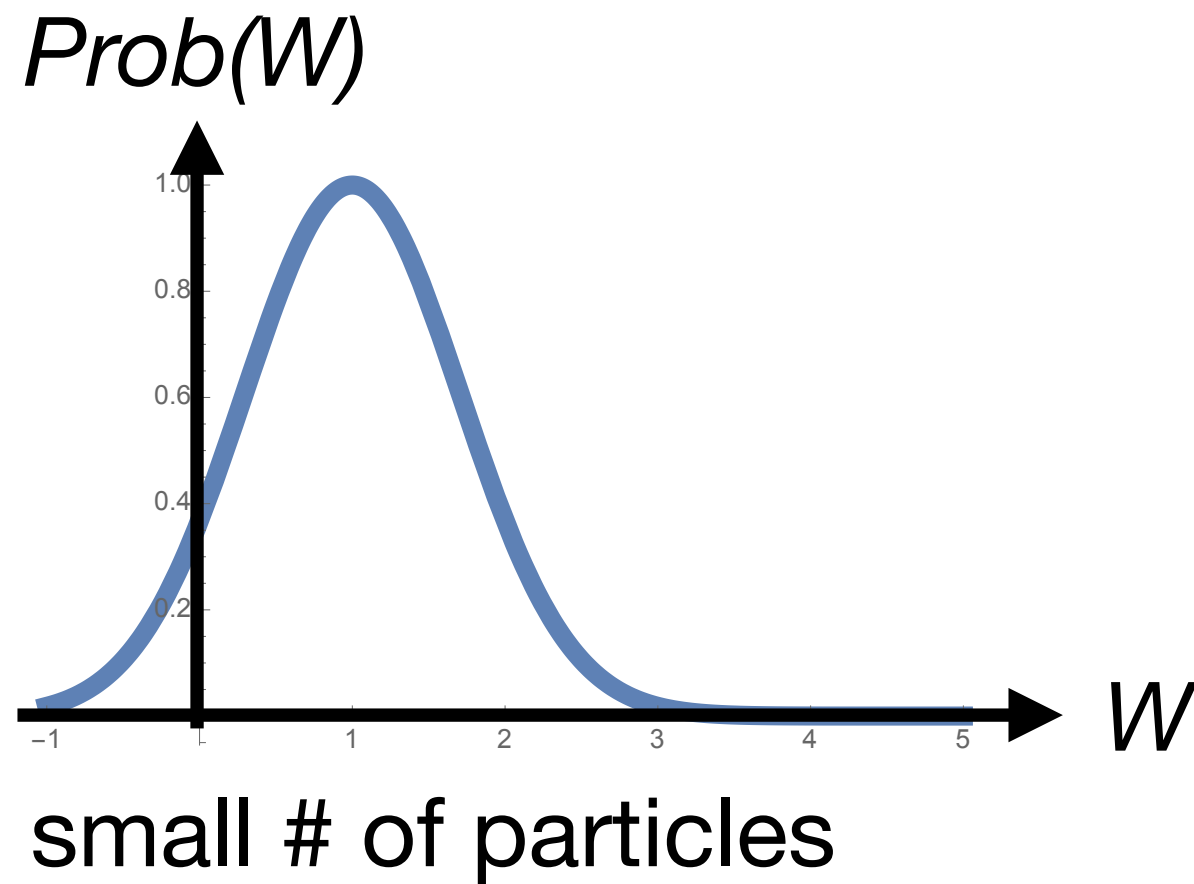


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Work is a **random variable** (for fixed process):

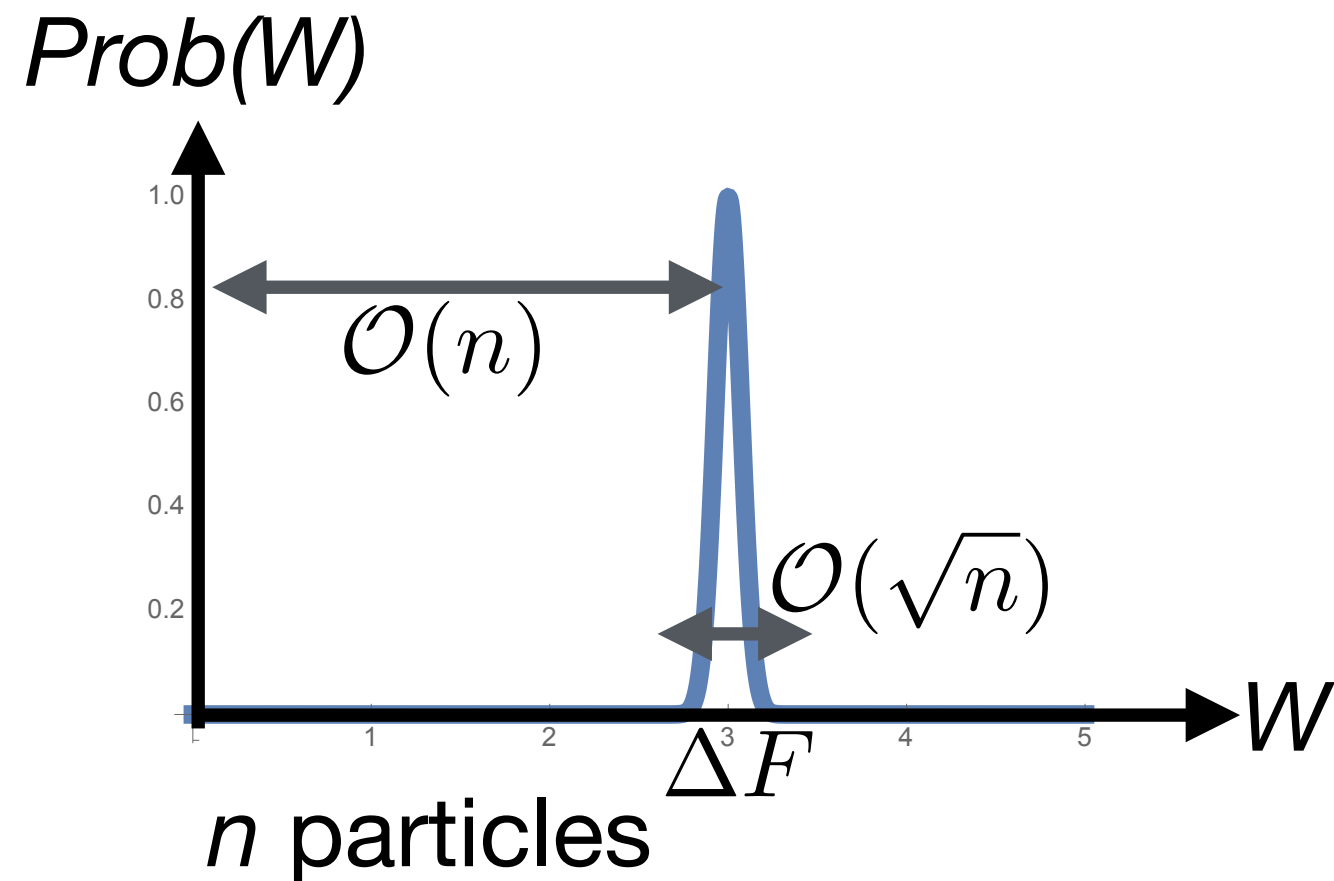
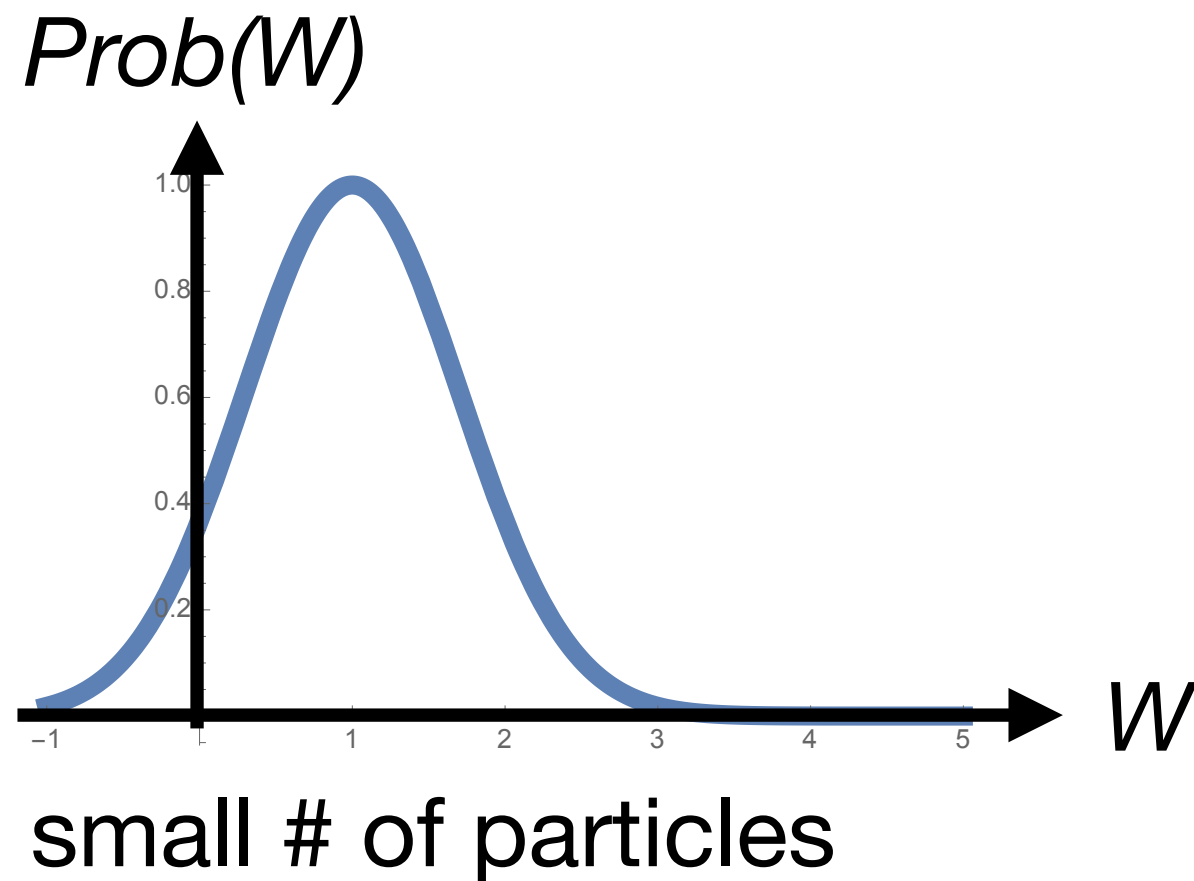
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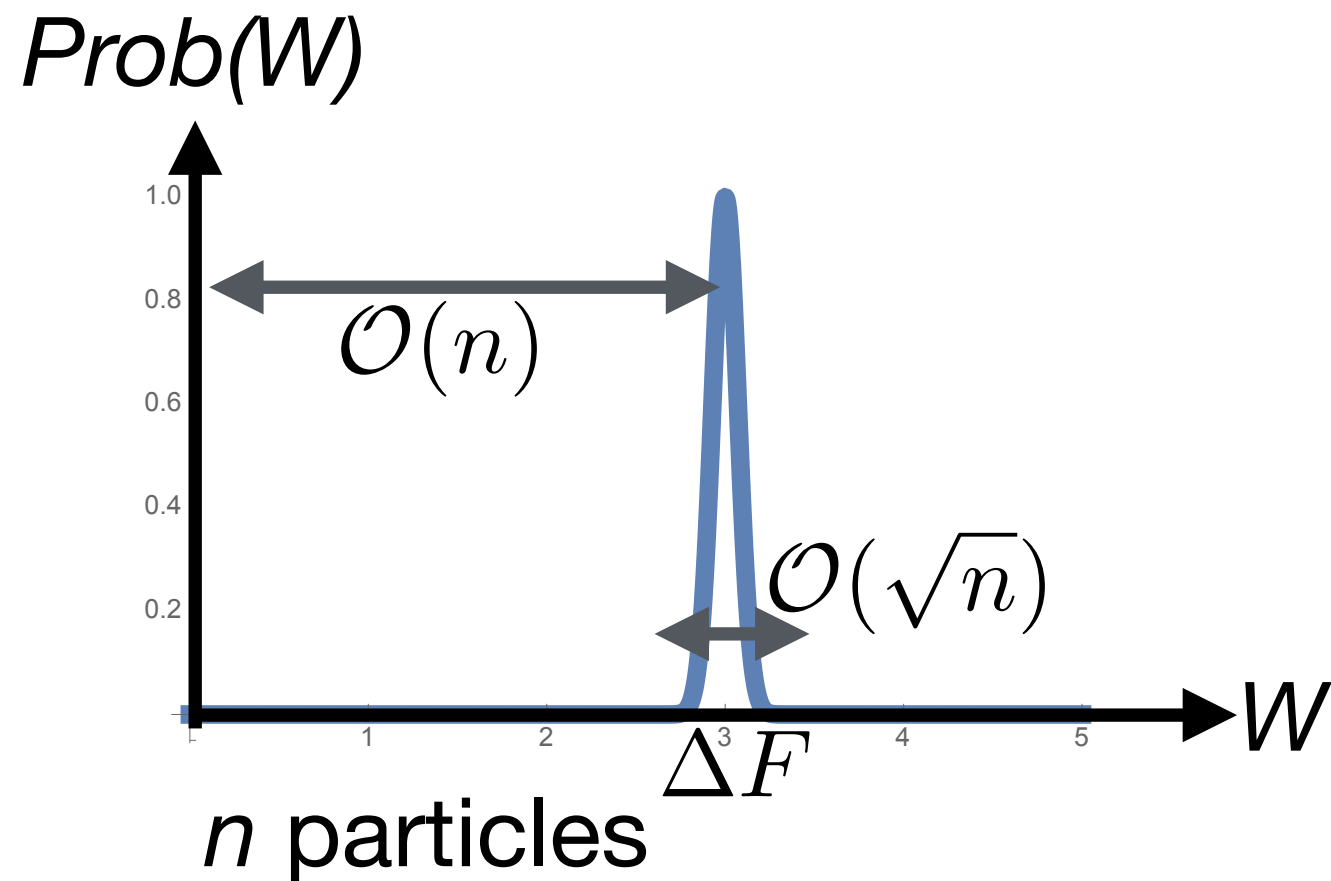
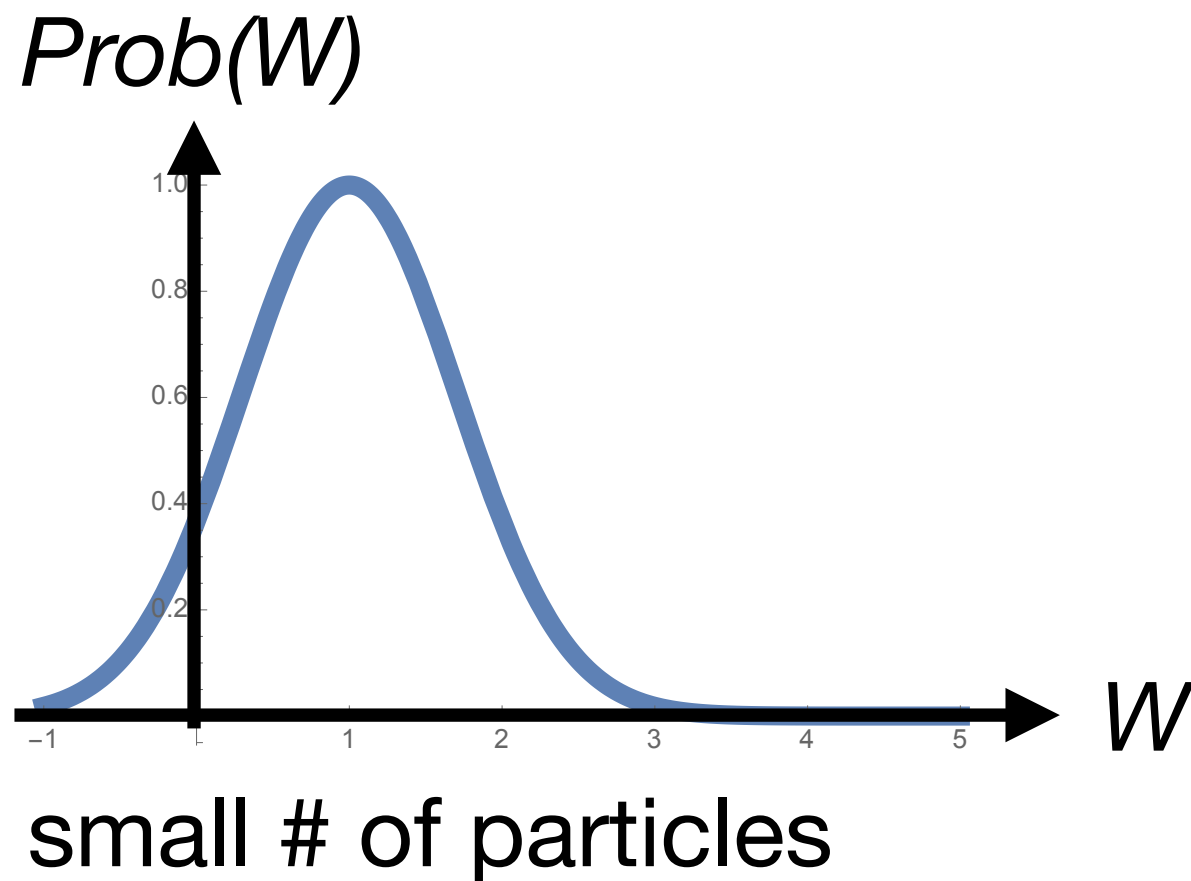


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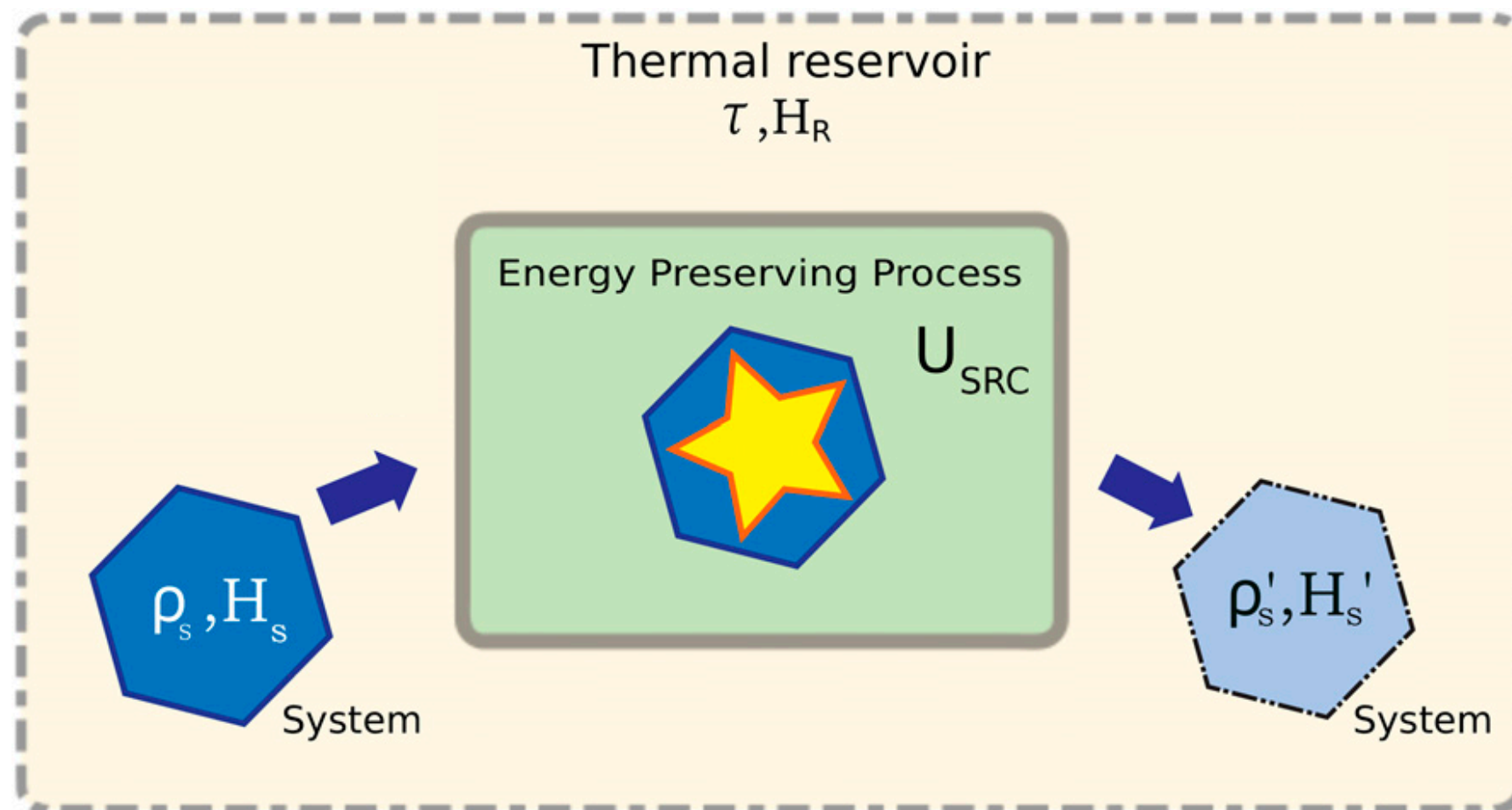
What can we say for “small” or strongly correlated systems?  
Work  $\approx$  its fluctuations  $\rightarrow$  reliability?

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## The rules of the game:

- It is “free” to bring in any “bath”  $B$  in its thermal state  $\gamma_B = \exp(-H_B/(k_B T))$ ,
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**Def.:** A *thermal operation*  $\mathcal{T}$  is a map of the form

$$\mathcal{T}(\rho_A) = \text{Tr}_B \left[ U_{AB} (\rho_A \otimes \gamma_B) U_{AB}^\dagger \right]$$

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**Question:** Which transitions (work extraction etc.) are possible via thermal operations?

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**Theorem** (Horodecki, Oppenheim, Nat. Comm. 4 (2013)):

For **block-diagonal** states,  $\rho_A \mapsto \rho'_A$  is possible via some thermal operation iff  $\rho_A$  *thermo-majorizes*  $\rho'_A$ .

# Thermodynamics as a resource theory

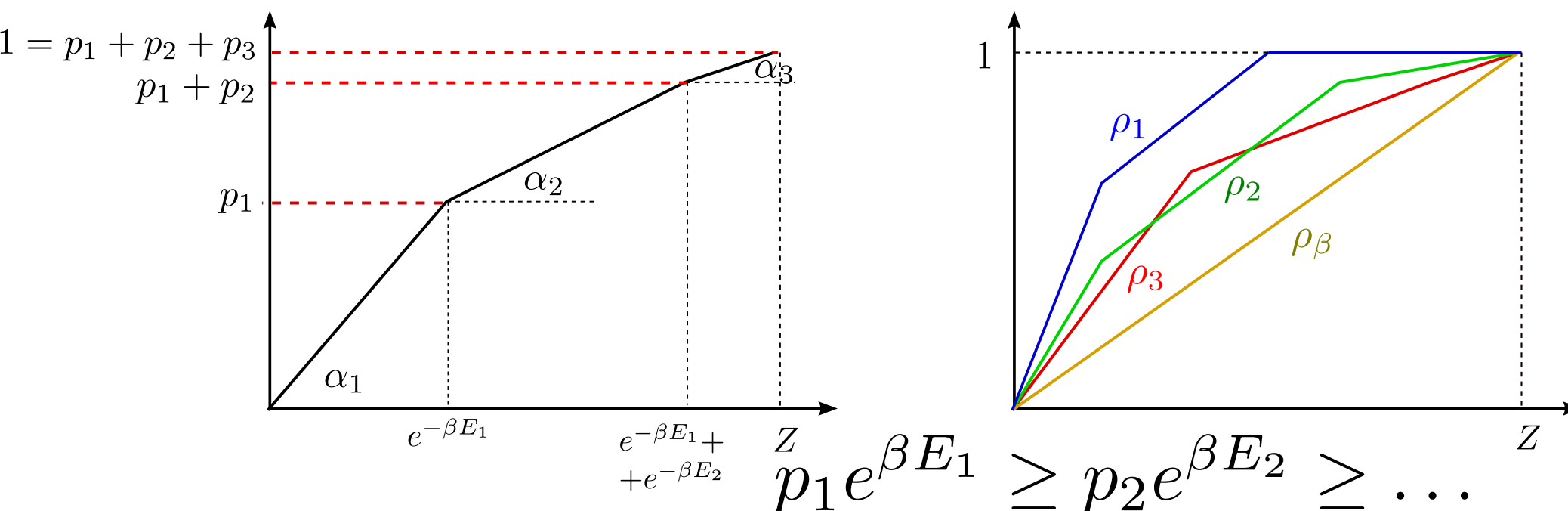
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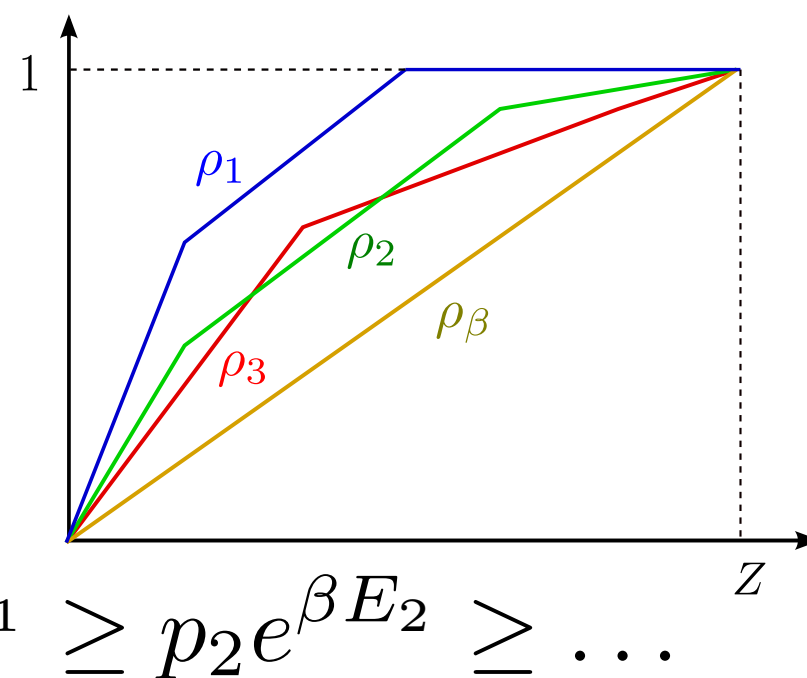
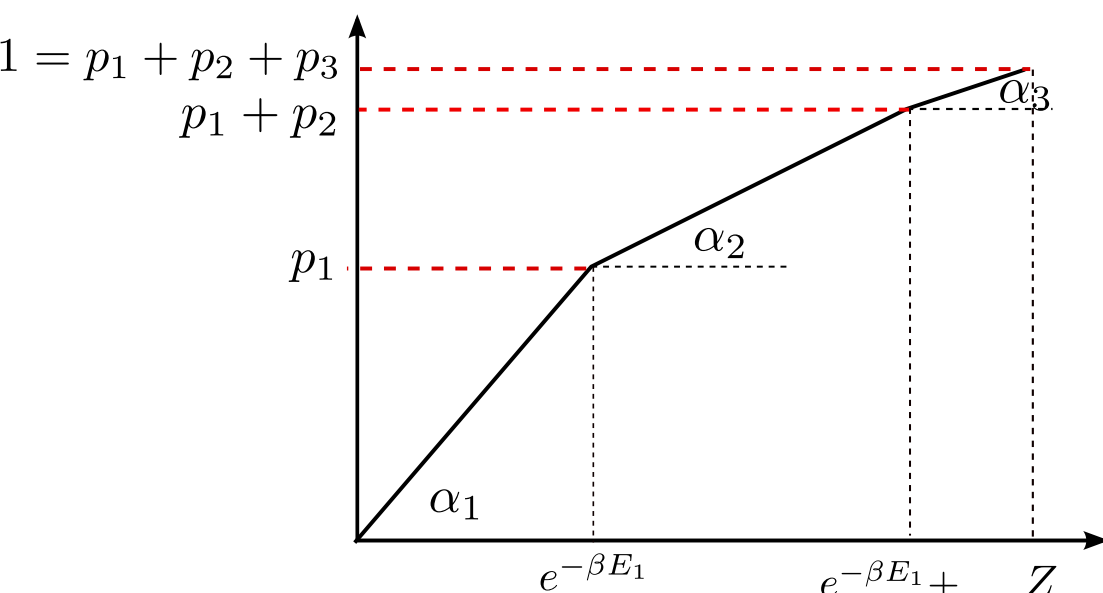
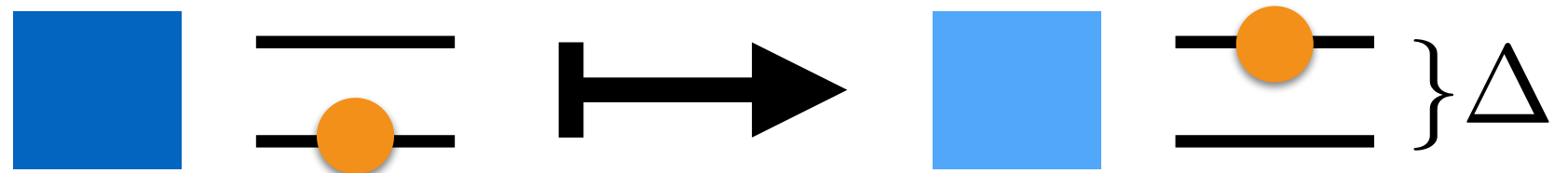
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# Work extraction and work of formation

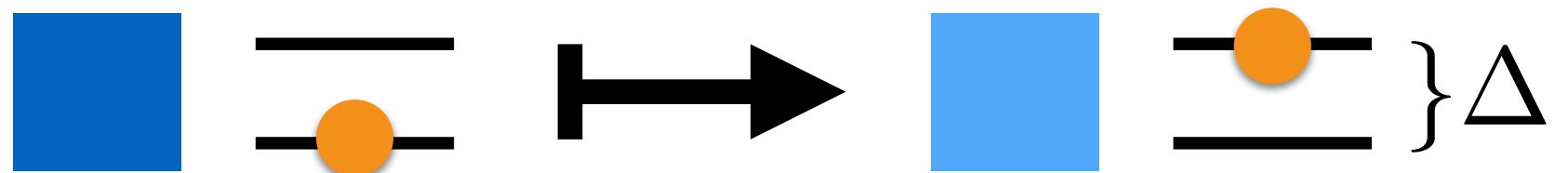
**Work extraction:**  $\sigma_A \otimes |g\rangle\langle g|_W \mapsto \sigma'_A \otimes |e\rangle\langle e|_W$



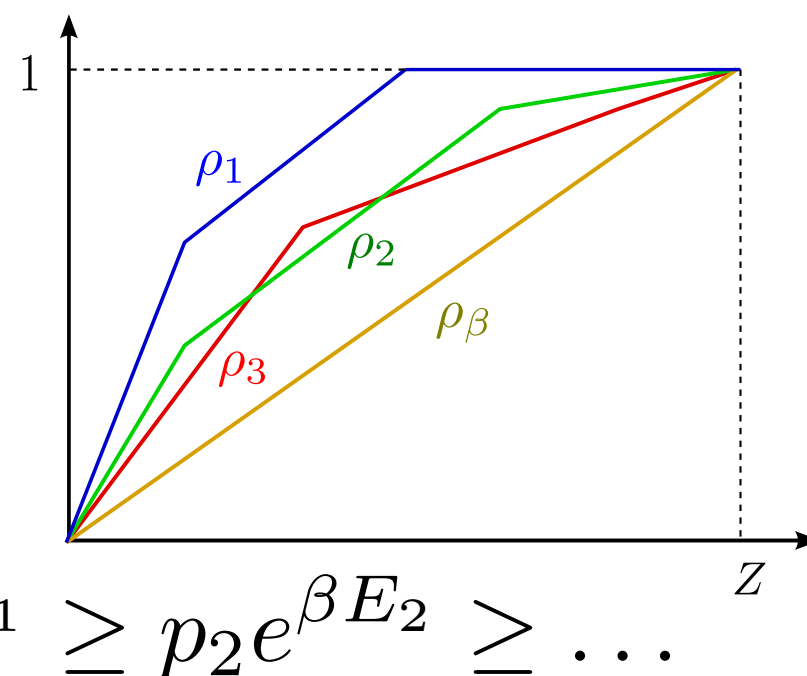
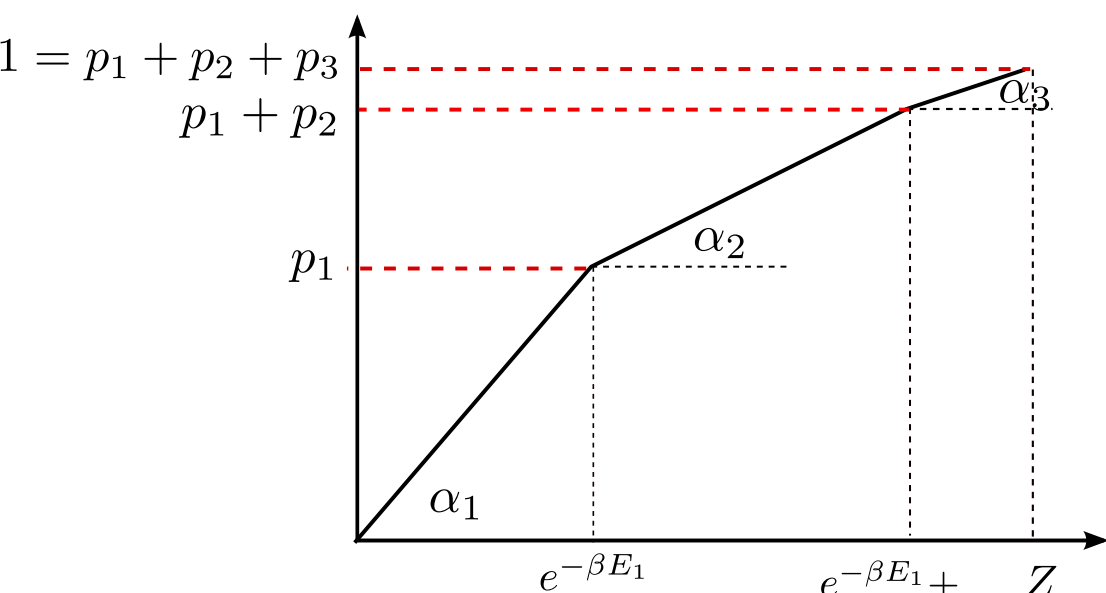
$$p_1 e^{\beta E_1} \geq p_2 e^{\beta E_2} \geq \dots$$

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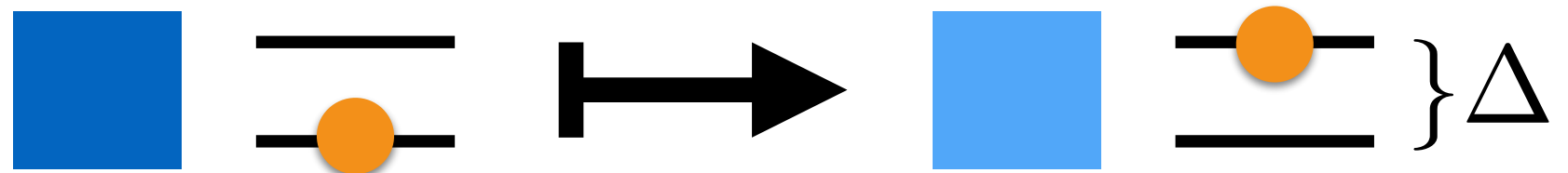


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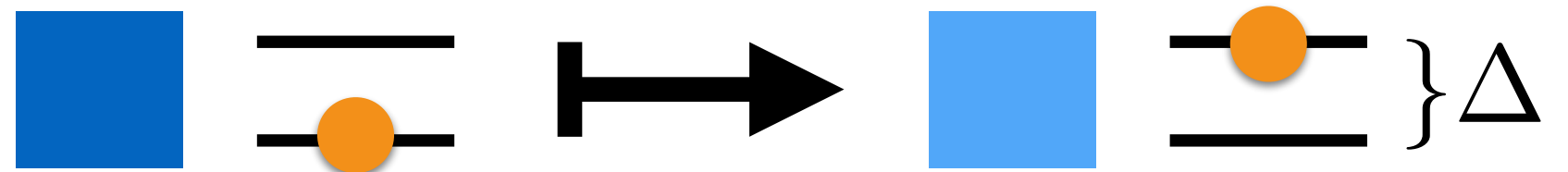
**Result:** *extractable work* is  $F_0(\sigma_A) - F(\gamma_A)$ ,

$$\text{where } F_0(\sigma) = k_B T \log \sum_{p_i \neq 0} e^{-\beta E_i}.$$

Similarly, *work cost* is  $F_\infty(\sigma_A) - F(\gamma_A)$   
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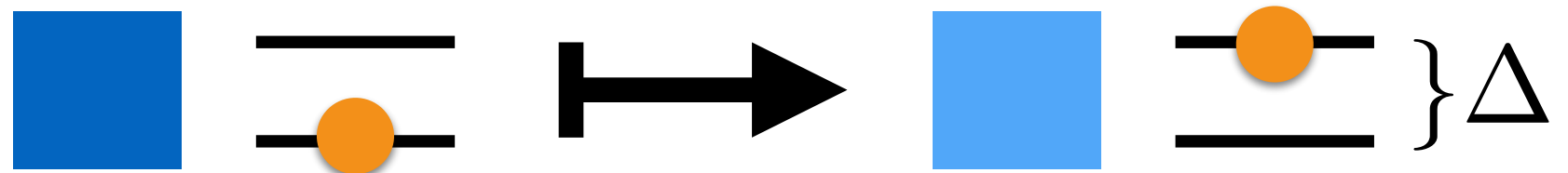
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**Fundamental irreversibility:**  $F_0 \ll F \ll F_\infty.$

## Thermodynamic limit

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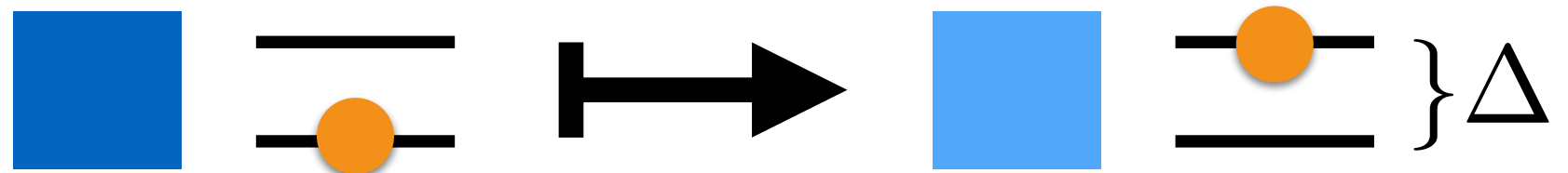
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Allowing small errors  $\varepsilon$ , we have

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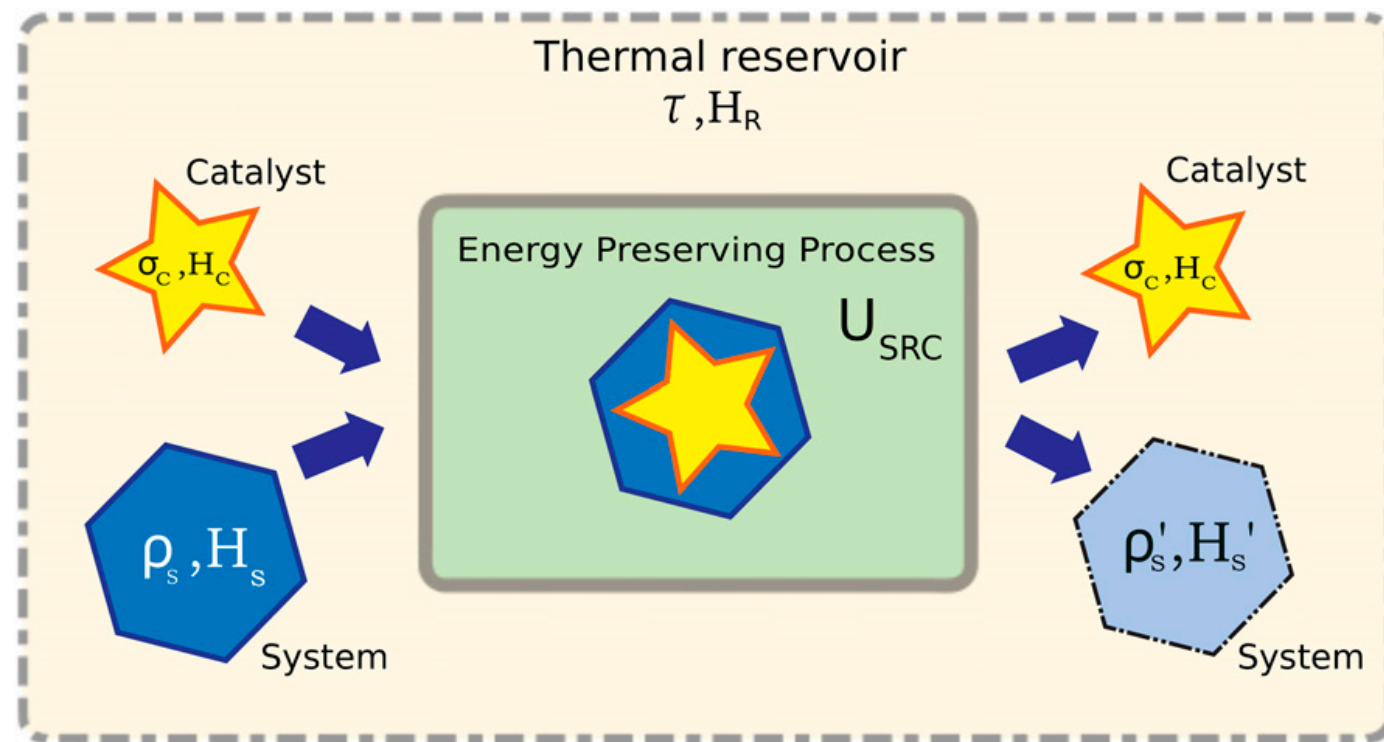
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(Rates of) work cost and extractable work become  $F$ .  
**Reversibility is restored** in the thermodynamic limit!

## General state transitions: **catalysts**

Allow for additional system  $C$  that is involved but doesn't change.

Brandão et al., *The second laws of quantum thermodynamics*, PNAS **112**, 3275 (2015).

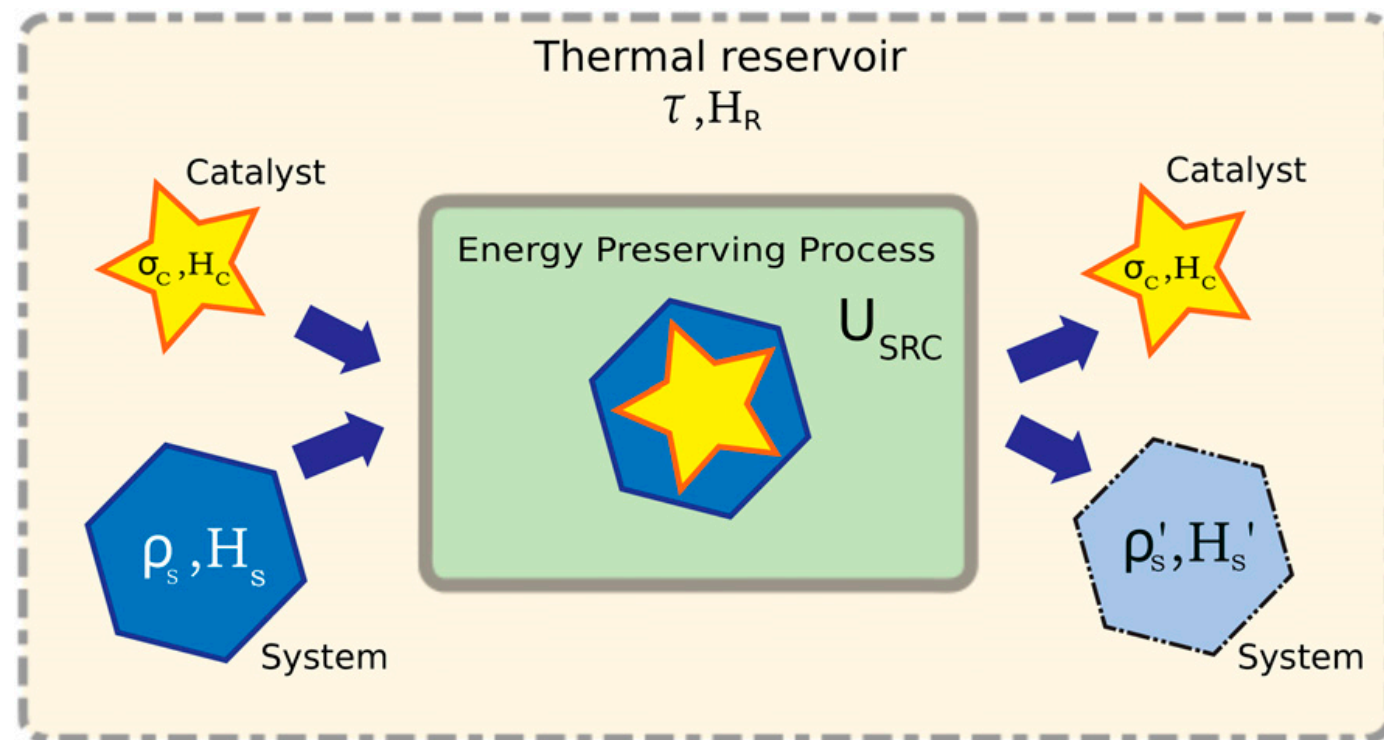


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$$[U_{SRC}, H_S + H_R + H_C] = 0$$

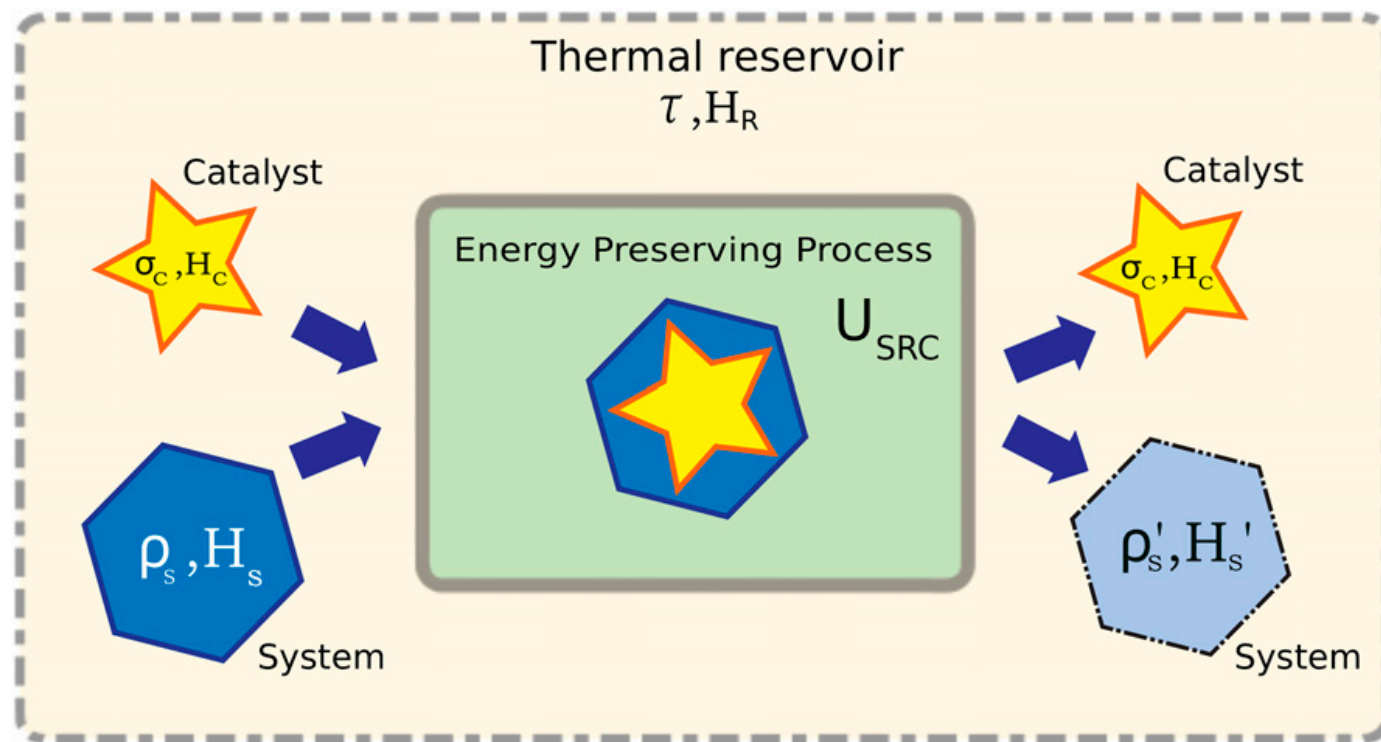
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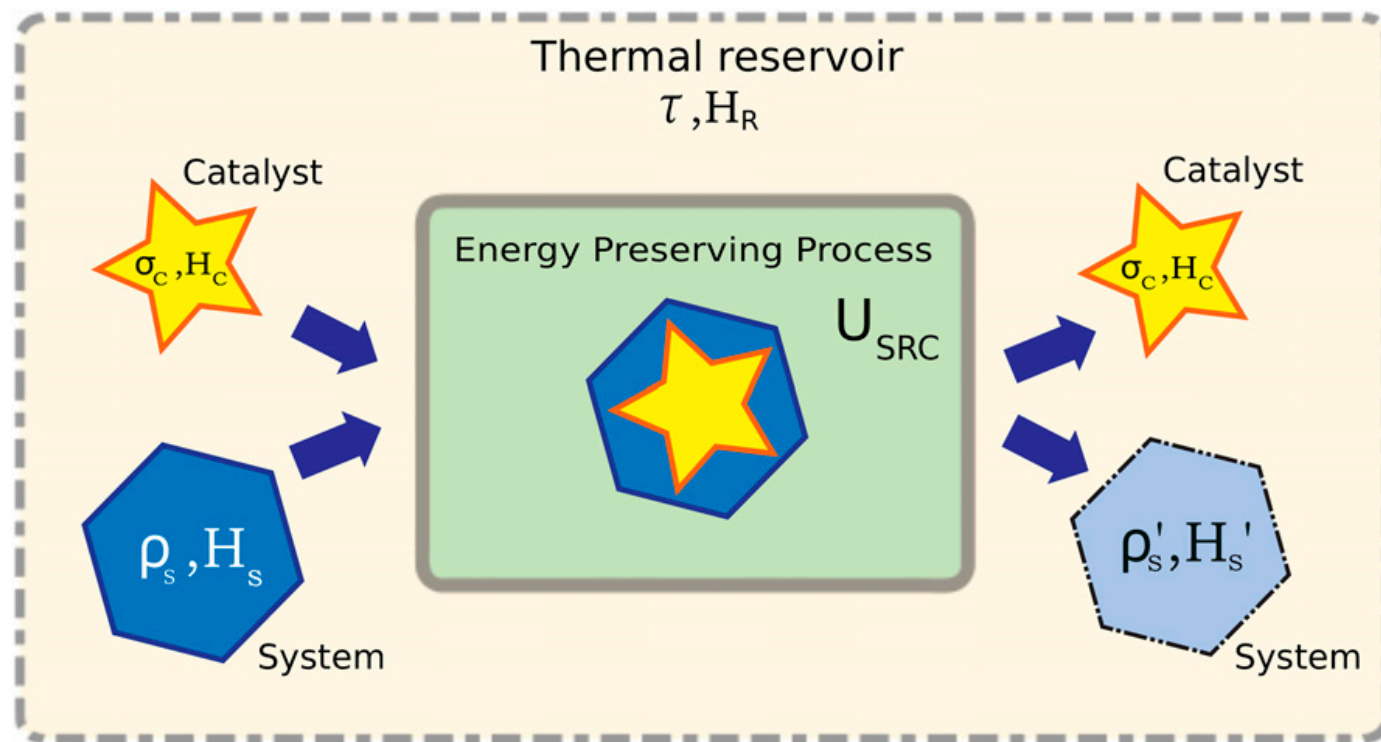
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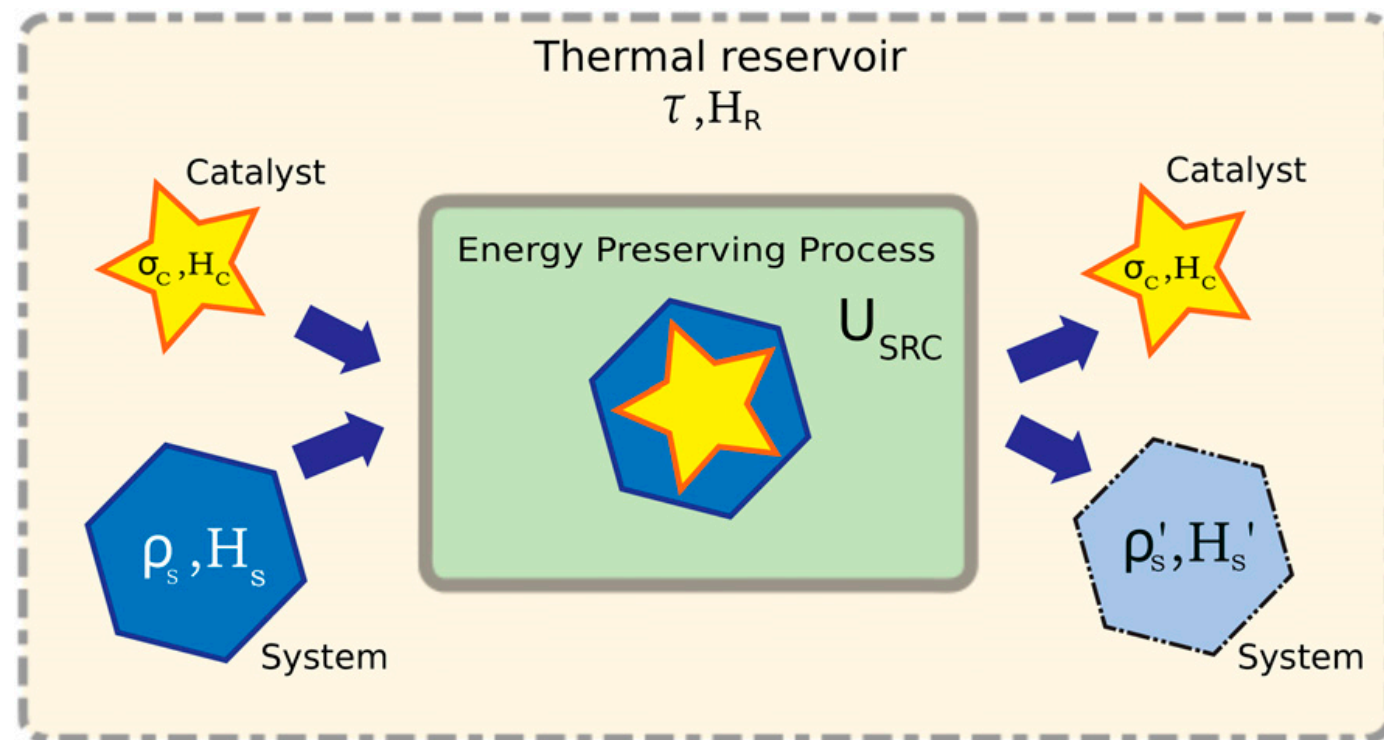
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**Theorem:** Possible if and only if  $F_\alpha(\rho_S) \geq F_\alpha(\rho'_S)$  for all  $\alpha \geq 0$ .

“Second laws” of thermodynamics. Note:  $F_{\alpha=1} = F$ .

# One-shot interpretation of non-equilibrium **F**



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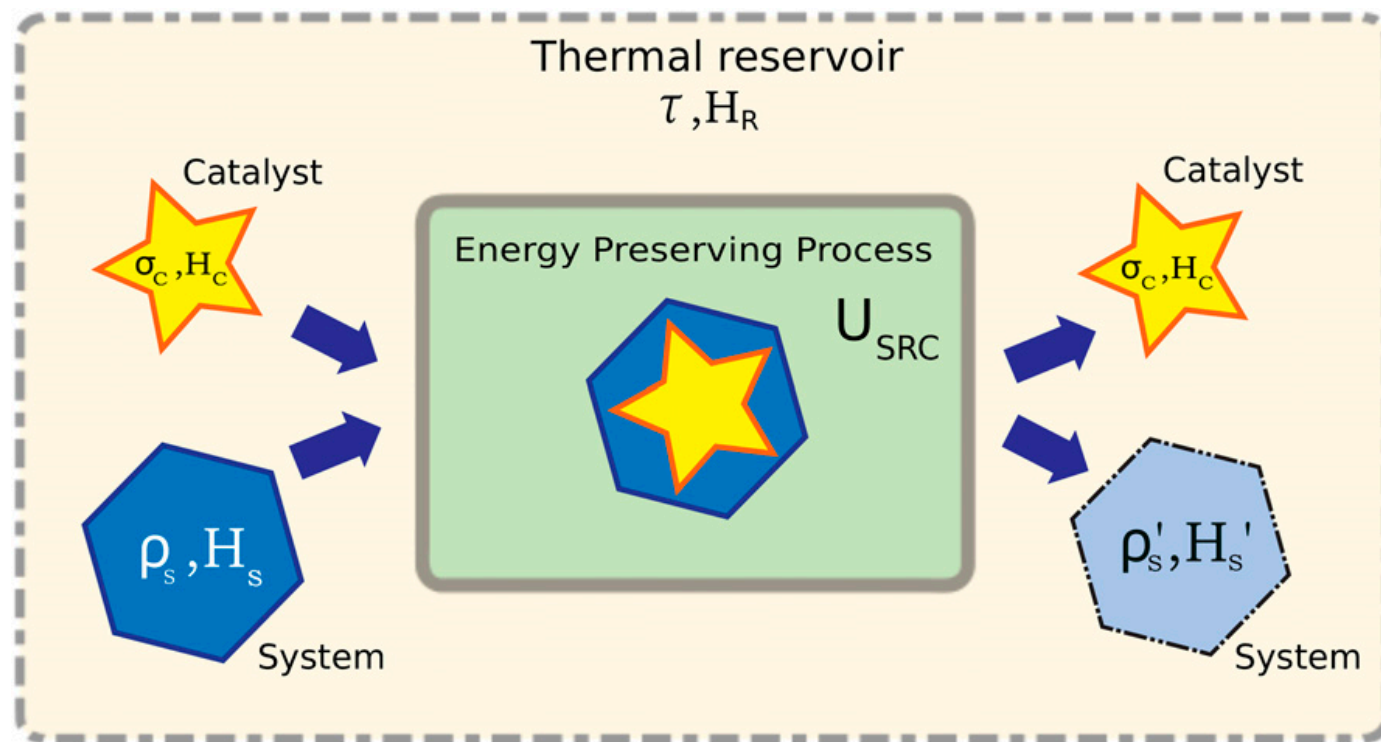
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Own work: **allow correlations** between catalyst and system.

[1] MM, Phys. Rev. X **8**, 041051 (2018).



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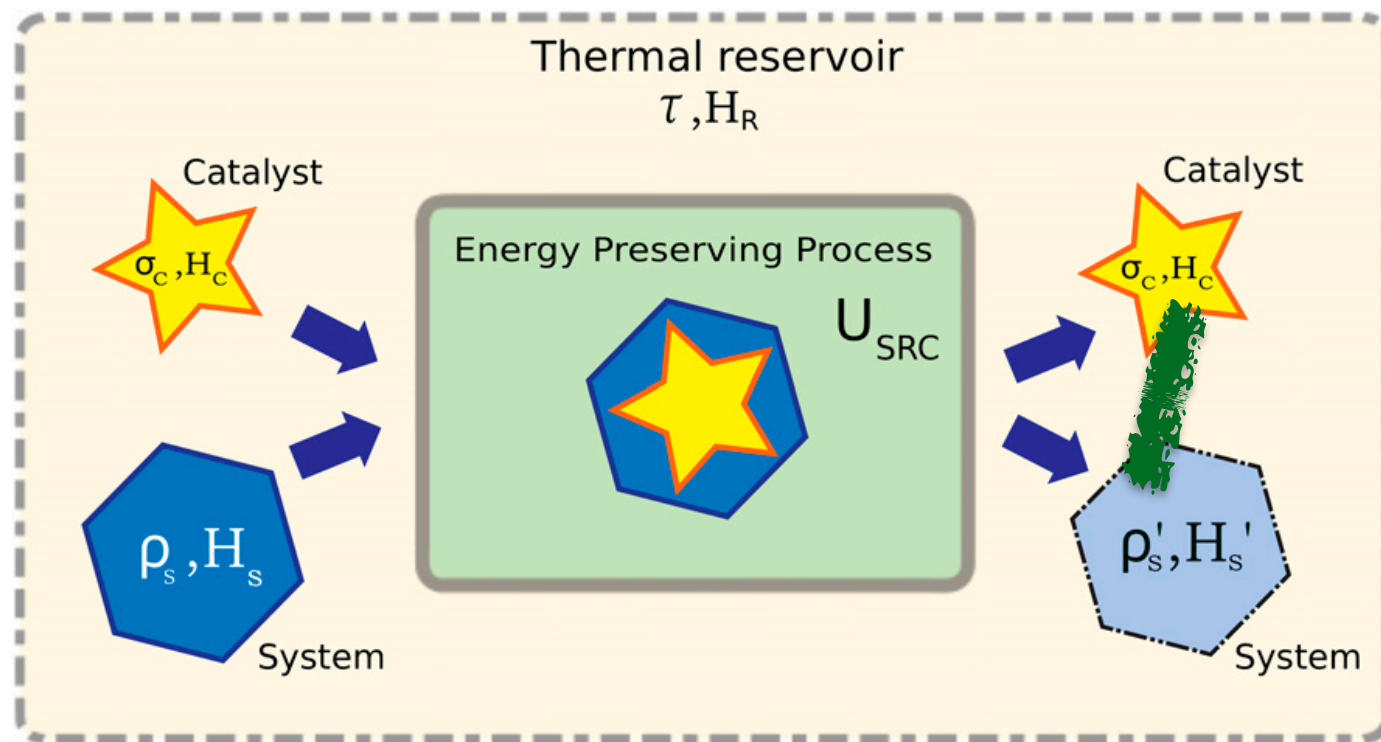
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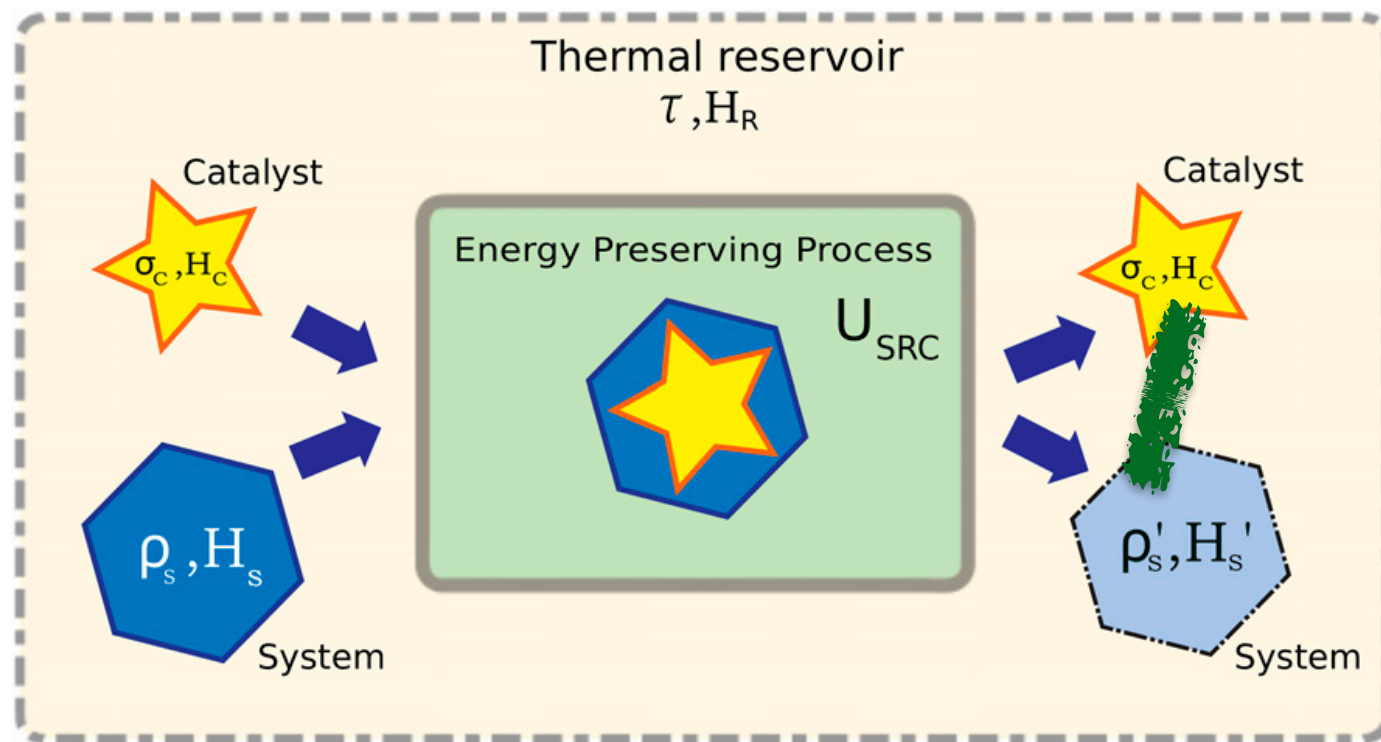
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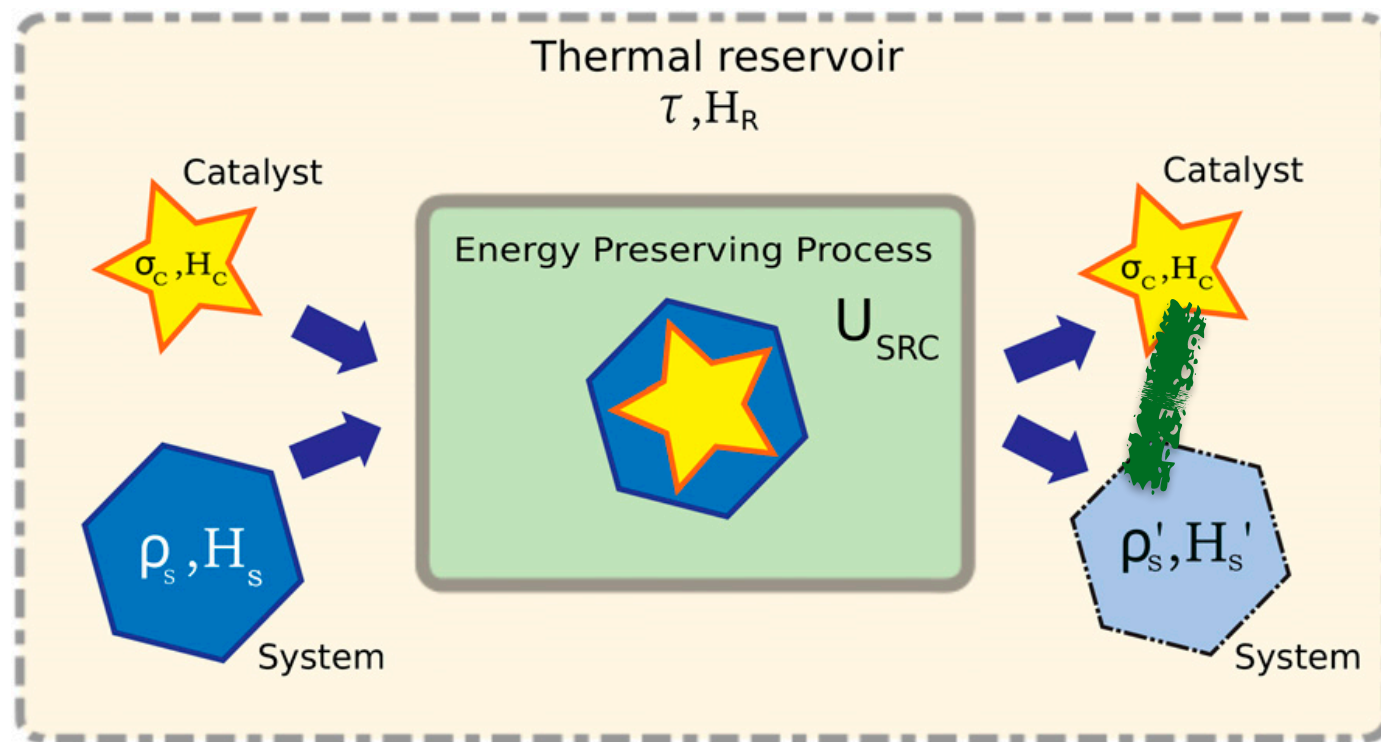
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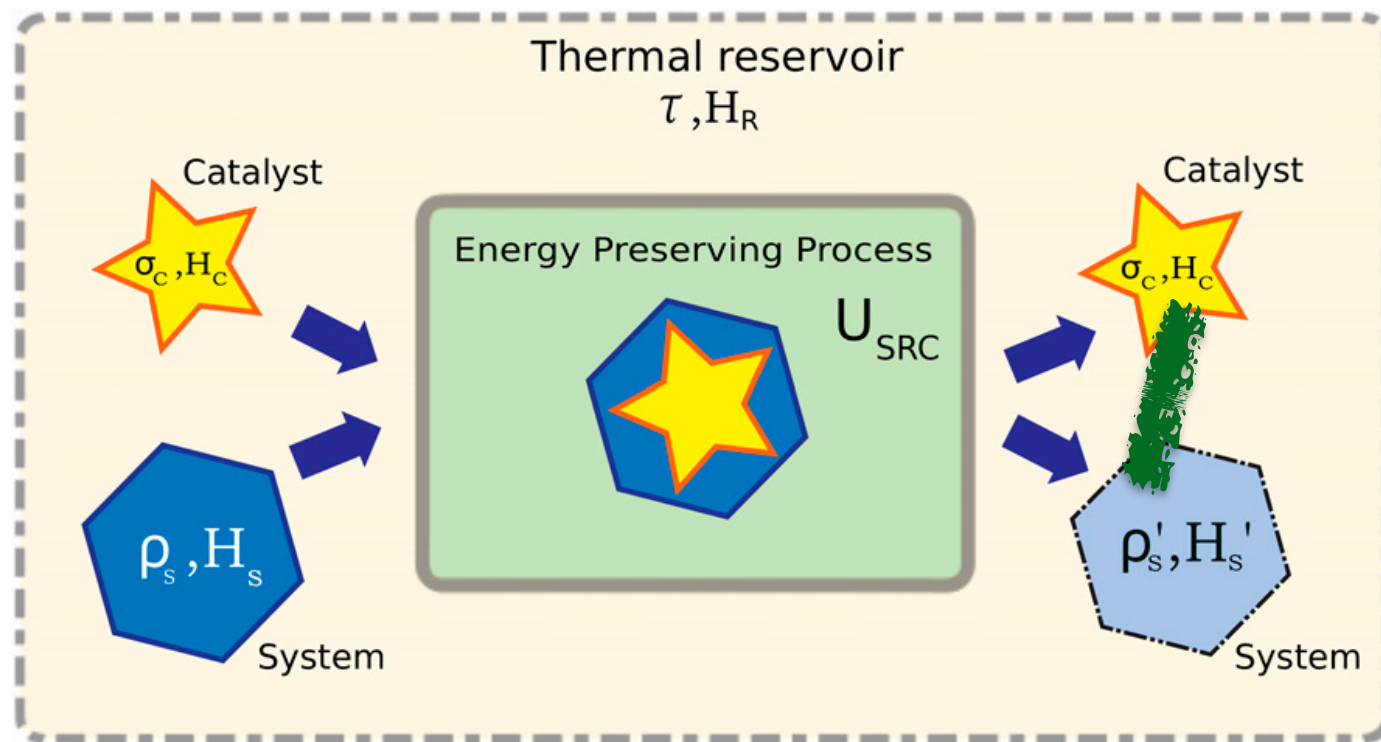
**Theorem [1]:** Possible if and only if  $F(\rho_S) \geq F(\rho'_S)$ .

**One-shot** interpretation of the free energy  $F$ .

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- Fluctuation-free **work cost**: If then transition possible while work bit  $|e\rangle_W \mapsto |g\rangle_W$ .
- *Almost* fluct.-free **work extraction**: If  $F(\rho_A) - F(\rho'_A) > \Delta > 0$ , then transition possible while work bit (for arbitrary  $\delta > 0$ )  
 $|g\rangle\langle g|_W \mapsto (1 - \delta)|e\rangle\langle e|_W + \delta \mathbf{1}/d$ .



## Questions

- Can the resource-theoretic approach be applied to “one-shot regime” of classical reversible computation?

E.g. if irreversibility cannot be completely avoided, can it at least be implemented in the “least costly” way?

- Can it be suitably modified to incorporate realistic constraints arising in classical reversible computation?

C. Perry, P. Ćwikliński, J. Anders, M. Horodecki, and J. Oppenheim, *A Sufficient Set of Experimentally Implementable Thermal Operations for Small Systems*, Phys. Rev. X **8**, 041049 (2018).

**Thank you!**